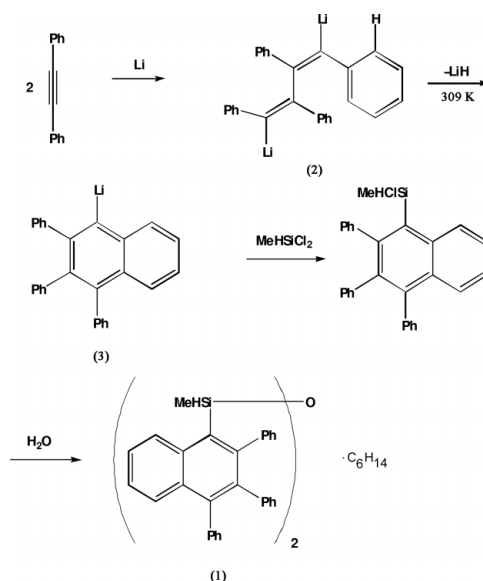


1,3-Dimethyl-1,3-bis(2,3,4-triphenyl-1-naphthyl)disiloxane hexane solvate

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bolte@chemie.uni-frankfurt.deThe title compound, C₅₈H₄₆OSi₂·C₆H₁₄, was formed in small yield as a by-product during the synthesis of 1-methyl-2,3,4,5-tetraphenylsilole.Received 20 October 2003
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Comment

The title compound, (1), was formed in small yields (5–10%) as a by-product during the synthesis of 1-methyl-2,3,4,5-tetraphenylsilole (Schuppan, 1998; Schuppan *et al.*, 2001). A possible reaction path for the formation of (1) is shown in the scheme below.

Key indicators

Single-crystal X-ray study

T = 173 K

Mean $\sigma(\text{C}-\text{C}) = 0.009 \text{ \AA}$

Some non-H atoms missing

Disorder in solvent or counterion

R factor = 0.134

wR factor = 0.363

Data-to-parameter ratio = 26.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The key step for its formation is LiH elimination from 1,4-dilithio-1,2,3,4-tetraphenylbutadiene, (2), at elevated temperature, yielding 1-lithio-2,3,4-triphenyl-1-naphthalene, (3). Subsequent reaction with methylchlorosilane and hydrolysis results in the formation of (1). Separation from the major product was achieved by column chromatography (hexanes–ethyl acetate = 20:1; SiO₂). Recrystallization from toluene gives colourless crystals suitable for X-ray crystallography. A search of the Cambridge Structural Database (CSD, Version 5.24 with two updates, April 2003; Allen, 2002) finds 438 structures containing the Si–O–Si fragment with four-coordinated Si and two-coordinated O atoms. Only 13 of these show an Si–O–Si angle less than 133°. This means that the angle observed in (1), 132.9 (3)°, is at the lower end of the range. The two naphthalene moieties enclose an angle of 27.07 (17)°. The dihedral angles between the naphthalene moiety C1–C10 and the attached phenyl rings are 70.69 (16), 69.31 (16) and 80.68 (15)° for C21–C26, C31–C36 and C41–

C46, respectively. For the second naphthalene moiety (C11–C20), the following dihedral angles were found: 83.39 (13)° for C51–C56, 74.53 (15)° for C61–C66, and 64.2 (2)° for C71–C76.

Experimental

Spectroscopic data: ^1H NMR [250 MHz, acetone [D_6], 300 K, $\delta(\text{CD}_3\text{COCD}_2\text{H}) = 2.1$]: δ 0.38 [d , $^3J(\text{HH}) = 3.5$ Hz, 6H], 4.28 [q , $^3J(\text{HH}) = 3.5$ Hz, 2H], 6.4 (m , 2H), 6.55–7.35 (m , 32H), 7.50 (m , 2H), 8.25 (m , 2H); ^{13}C NMR [62.9 MHz, CDCl_3 , 300 K, $\delta(\text{CDCl}_3) = 77.0$]: δ 1.3 (CH_3), 125.2 (CH), 125.6 (CH), 125.7 (CH), 126.2 (CH), 126.4 (CH), 126.5 (CH), 126.8 (CH), 126.9 (CH), 127.3 (CH), 127.6 (CH), 127.8 (CH), 128.0 (CH), 128.6 (CH), 130.3 (CH), 130.8 (CH), 130.8 (CH), 131.1 (CH), 131.3 (CH), 131.3 (CH), 131.8 (C^q), 133.5 (C^q), 136.2 (C^q), 138.2 (C^q), 139.6 (C^q), 140.2 (C^q), 140.6 (C^q), 142.3 (C^q), 147.5 (C^q); ^{29}Si NMR [79.5 MHz, CDCl_3 , 300 K, $\delta(\text{TMS}) = 0.0$]: δ 14.5.

Crystal data

$\text{C}_{58}\text{H}_{46}\text{OSi}_2\cdot\text{C}_6\text{H}_{14}$	$D_x = 1.172 \text{ Mg m}^{-3}$
$M_r = 901.30$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 3857 reflections
$a = 18.716$ (4) Å	$\theta = 1\text{--}25^\circ$
$b = 15.533$ (3) Å	$\mu = 0.11 \text{ mm}^{-1}$
$c = 17.685$ (4) Å	$T = 173$ (2) K
$\beta = 96.67$ (2)°	Plate, colourless
$V = 5106.5$ (19) Å ³	$0.30 \times 0.30 \times 0.08 \text{ mm}$
$Z = 4$	

Data collection

Siemens SMART CCD three-circle diffractometer	14554 independent reflections
ω scans	4271 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.126$
$T_{\text{min}} = 0.967$, $T_{\text{max}} = 0.991$	$\theta_{\text{max}} = 30.7^\circ$
77354 measured reflections	$h = -25 \rightarrow 25$
	$k = -21 \rightarrow 21$
	$l = -24 \rightarrow 24$

Refinement

Refinement on F^2	H atoms treated by a mixture of independent and constrained refinement
$R[F^2 > 2\sigma(F^2)] = 0.134$	$w = 1/[\sigma^2(F_o^2) + (0.146P)^2]$
$wR(F^2) = 0.363$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 0.97$	$(\Delta/\sigma)_{\text{max}} < 0.001$
14554 reflections	$\Delta\rho_{\text{max}} = 0.50 \text{ e \AA}^{-3}$
558 parameters	$\Delta\rho_{\text{min}} = -0.37 \text{ e \AA}^{-3}$

Table 1

Selected geometric parameters (Å, °).

Si1–O1	1.644 (4)	O1–Si2	1.637 (4)
Si2–O1–Si1	132.9 (3)		

All H atoms were located by difference Fourier synthesis and refined with fixed individual displacement parameters [$U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$] using a riding model with aromatic C–H = 0.95 Å and methyl C–H = 0.98 Å. The H atoms bonded to Si were refined isotropically with an Si–H distance restrained to

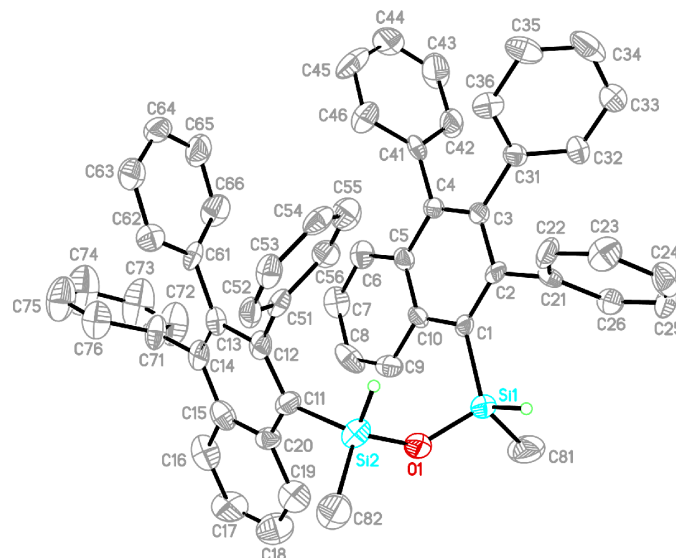


Figure 1

A perspective view of the title compound with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

1.45 (6) Å. The hexane molecule is severely disordered. The contribution of the disordered solvent to the calculated structure factors was taken into account with the *BYPASS* algorithm (van der Sluis & Spek, 1990), implemented as the *SQUEEZE* option in *PLATON* (Spek, 2003). The final F_o^2/F_c^2 data were calculated with the FCF routine of *PLATON* and include the disordered solvent contribution. The inclusion of such a large percentage of weak data in the refinement is probably responsible for the high R factors. This, in turn, affects the precision of the results. The crystal did not appear to be twinned.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *PLATON* (Spek, 2003).

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